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# <sup>73</sup>Ge NMR spectra of 1,3-dioxa-6-aza-2-germacyclooctanes

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#### Abstract

The spirocyclic derivatives of 1,3-dioxa-6-aza-2-germacyclooctanes (II) have been studied by <sup>73</sup>Ge NMR spectroscopy. Evidence for the existence in II of a transannular donor-acceptor  $N \rightarrow Ge$  bond is presented.

Until recently the only derivatives of pentacoordinate Ge studied by <sup>73</sup>Ge NMR were the germatranes ROGe(OCH<sub>2</sub>CH<sub>2</sub>)<sub>3</sub>N (I) containing an intramolecular donor-acceptor (DA) N $\rightarrow$ Ge bond [1,2]. Here we report <sup>73</sup>Ge NMR study of the bicyclic analogues of I, i.e. 1,3-dioxa-6-aza-2-germacyclooctanes (germocanes, II).

$$R \xrightarrow{O}_{O} Ge \xrightarrow{O}_{O} X = (CH_{2})_{2};$$

$$(I: X = (CH_{2})_{2};$$

$$2: X = (CMe_{2})_{2};$$

$$3: X = o-C_{6}H_{4};$$

$$4: X = (CH_{2}CH_{2})_{2}O) \left\{ \begin{array}{c} a: R = H \\ b: R = Me \\ c: R = Pr^{i} \end{array} \right\}$$
(II)

An increase in <sup>73</sup>Ge shielding in derivatives of II (see Table 1), as compared to that observed for the model compound  $Ge(OEt)_4$  [3], provides evidence for the existence of a transannular DA N  $\rightarrow$  Ge bond in II in solution. This agrees well with the results of an X-ray diffraction study of crystalline 1b in which the distance between the N and Ge atoms is decreased to 2.159 Å [4]. Substitution of the H atom at N by a Me or Pr<sup>i</sup> group does not appreciably affect the <sup>73</sup>Ge shielding and, consequently the DA N  $\rightarrow$  Ge bond strength.

The insignificant upfield shift of the <sup>73</sup>Ge resonance in **2b** compared with that of **1b** is attributable to the inductive effect of the Me groups in the  $\beta$ -position which is consistent with the effects observed in tetraalkoxygermanes [3]. A similar effect is observed for compound **3b**, but the lack of <sup>73</sup>Ge data for the appropriate model compounds prevents definite conclusions to be drawn.

compounds <sup>a</sup>	oxa-6-aza-2-germacyclooctanes (11)	) in CD <sub>3</sub> CN at 343	K and for	model
Compound	δ <sup>73</sup> Ge	$\Delta v_{1/2}$		
1a	- 57	280		

320

500

170

150

240

30

165

- 55

- 56

- 57

- 82

- 94

- 44

-68

73 -

<sup>a</sup> Chemical shifts ( $\delta$ ) in ppm relative to Me<sub>4</sub>Ge, linewidth at half-height ( $\Delta \nu_{1/2}$ ) in Hz. <sup>73</sup>Ge NMR spectra were recorded on a Bruker WM-360 spectrometer at 12.56 MHz in 5-mm sample tubes. <sup>b</sup> In DMSO-d<sub>6</sub>. <sup>c</sup> Ref. 3. <sup>d</sup> In CDCl<sub>3</sub>, ref. 1.

Decreased <sup>73</sup>Ge shielding in the series 1 and 2, as compared to the alkoxy derivatives of I ( $\delta^{73}$ Ge = -60 to -68 ppm [1,2]), suggests a stronger DA N  $\rightarrow$  Ge bond in compounds I. However, the differences in the conditions under which <sup>73</sup>Ge spectra were recorded must be taken into account.

A substantial increase in  $^{73}$ Ge shielding in compound 4a is indicative of a pseudohexacoordinate Ge resulting from additional 1,5-transannular  $O \rightarrow Ge$  bonding. At present, unambiguous interpretation of the results obtained is not possible because of insufficient information regarding the influence of ring size on the  $^{73}$ Ge chemical shifts.

The strong temperature dependence of the <sup>73</sup>Ge linewidth is also indicative of the existence of DA  $N \rightarrow Ge$  bonding in II. For most of the II derivatives the linewidth at half-height of the <sup>73</sup>Ge resonance was ca. 200 Hz (see Table 1) and was found to rise with fall in temperature. Only compound 4a gave <sup>73</sup>Ge signals at ambient temperature in DMSO- $d_6$  solution. The enhanced asymmetry of the charge distribution around Ge in the pentacoordinate state results in an increase in quadrupolar relaxation rate of the <sup>73</sup>Ge nuclei, thus no <sup>73</sup>Ge signals were observed for the other compounds at ambient temperature in DMSO- $d_6$ .

Consequently, the <sup>73</sup>Ge NMR data favour the existence of an intramolecular  $N \rightarrow Ge$  coordination bond in 1,3-dioxa-6-aza-2-germacyclooctanes in solution.

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1b 1c

2b

3Ь

4a b

Ge(OEt), °

N(CH<sub>2</sub>CH<sub>2</sub>O)<sub>3</sub>GeOEt <sup>d</sup>

Table 1